

Structure of Na₂O-GeO₂-P₂O₅ glasses by X-ray and neutron diffraction

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The Ge-O coordination number (N_{GeO}) is 4 in vitreous GeO₂. The mixed network former GeO₂-P₂O₅ glasses show an increase of the Ge-O coordination number (N_{GeO}) with increasing P₂O₅ content due to the asymmetric distribution of bond valencies in the favored P-O-Ge bridges [1]. Large N_{GeO} 's with ~5 were obtained in K₂O-GeO₂-P₂O₅ glasses [2]. The origin of this N_{GeO} increase differs from that of the germanate anomaly where the absence of non-bridging oxygen is related to the increase of N_{GeO} .

The Na₂O-GeO₂-P₂O₅ glasses here investigated possess larger P₂O₅ contents than the samples used before [2]. The structural features of the samples are determined by X-ray and neutron diffraction experiments of excellent resolving power. The strong X-ray scattering of Ge and sufficiently different lengths of the P-O and Ge-O bonds allow to resolve the Ge-O peak and to extract N_{GeO} . The resulting values vary in a range $4.5 \leq N_{\text{GeO}} \leq \sim 6$. N_{GeO} depends on the P₂O₅ content. Its maximum value is obtained for the sample of only 20 mol% GeO₂.

The pre-peaks in the structure factors $S(Q)$ at position $Q_1 = \sim 8 \text{ nm}^{-1}$ that were reported earlier [2] appear also in all the $S(Q)$ of the present glasses. This pre-peak is related to a length of medium-range order of ~1.0 nm. The distance is explained with the mutual order of micro-domains of alkali-phosphate and Ge-phosphate regions [2]. The shape of the pre-peak and its position at Q_1 are widely independent of the GeO₂ content. This model gets support from other authors who reported a combined NMR and X-ray photoelectron spectroscopy study of GeO₂-NaPO₃ glasses [3]. The preferred attraction of Na⁺ ions by the phosphorus component and the favored P-O-Ge linkages agree with our model assumptions.

[1] U. Hoppe, R.K. Brow, B.C. Tischendorf, P. J v ri, and A.C. Hannon, *J. Phys.: Condens. Matter* **18**, 1847 (2006)

[2] U. Hoppe, G. Walter, R.K. Brow, N.P. Wyckoff, A. Sch ps, and A.C. Hannon, *Solid State Commun.* **143**, 403 (2007) and other papers

[3] J. Ren and H. Eckert, *J. Phys. Chem. C*, **116**, 12747 (2012)